

# Heles Year 12 Induction booklet

## OCR A – A Level Physics

### Foundations of A Level Physics

#### 1.1-2.2

Name:

Bring this workbook to every lesson

**Note – You will need to create a quizlet account with your Heles School email address if you don't already have one**

**Retrieval Questions – Available as Quizlet sets on classroom (scan QR code for direct access)**

**You are to screen shot your scores for each of these as they are checked**

#### 1 Prefixes



What is the value of the prefix T (tera)	$10^{12}$
What is the value of the prefix G (giga)	$10^9$
What is the value of the prefix M (mega)	$10^6$
What is the value of the prefix k (kilo)	$10^3$
What is the value of the prefix d (deci)	$10^{-1}$
What is the value of the prefix c (centi)	$10^{-2}$
What is the value of the prefix m (milli)	$10^{-3}$
What is the value of the prefix $\mu$ (micro)	$10^{-6}$
What is the value of the prefix n (nano)	$10^{-9}$
What is the value of the prefix p (pico)	$10^{-12}$

## 2.1-2.2 Quantities, Units and Conversions



What six base units of the SI system are used in A Level Physics?	metre, kilogram, second, Ampere, Kelvin, Mole
What six base quantities of the SI system are used in A Level Physics?	length, mass, time, electric current, absolute temperature, amount of substance.
What is the base unit of mass?	kg
What is the base unit of length?	m
What is the base unit of time?	s
What is the base unit of current?	A
What is the base unit of temperature?	K
What is the SI unit of potential difference?	V
What are the SI units of density?	$\text{kgm}^{-3}$
What is the SI unit of energy?	J
What is the SI unit of power?	W
What is the SI unit of charge?	C
What is $1\text{m}^2$ in $\text{mm}^2$ ?	$10^6\text{mm}^2$
What is $1\text{m}^3$ in $\text{cm}^3$ ?	$10^6\text{cm}^3$
What is $1\text{km}^2$ in $\text{m}^2$ ?	$10^6\text{m}^2$
What is $1\text{cm}^3$ in $\text{mm}^3$ ?	$10^3\text{mm}^3$
There are $10^{24}$ free charge carriers per $\text{m}^3$ of a material. How many are there per $\text{mm}^3$ ?	$10^{15}$
$10^6$ particles strike each $\text{mm}^2$ of a surface per second. How many strike $1\text{m}^2$ of the surface per second?	$10^{12}$
If there is 1 particle per $\text{mm}^3$ , how many are there per $\text{m}^3$ ?	$10^9$
Water has a density of $1000\text{kgm}^{-3}$ . What is this in $\text{kgcm}^{-3}$ ?	$10^{-3}$
Water has a density of $1000\text{kgm}^{-3}$ . What is this in $\text{gcm}^{-3}$ ?	1
What six base units of the SI system are used in A Level Physics?	metre, kilogram, second, Ampere, Kelvin, Mole

## Conversion of units/changes of magnitude of units

### Key ideas:

A – Converting areas and volumes can be tricky and often leads to mistakes by students. Why is this?

B – Physics deals with objects ranging in size from sub-atomic particles to whole galaxies. Using internet research or the Scale of the Universe website ( <https://scaleofuniverse.com/en-gb> ) list two objects that are in the following size ranges:

#### 1pm – 1nm

- 1.
- 2.

#### 1nm – 1 $\mu$ m

- 1.
- 2.

#### 1 $\mu$ m – 1mm

- 1.
- 2.

#### 1mm – 1m

- 1.
- 2.

#### 1m – 1km

- 1.
- 2.

#### 1km – 1Mm

- 1.
- 2.

#### 1Mm – 1Gm

- 1.
- 2.

#### 1Gm – 1Tm

- 1.
- 2.

#### 1Tm – 1Pm

1.

2.

## Exercises

### 1. Prefixes

Prefix	Symbol	Name	Multiplier
femto	f	quadrillionth	$10^{-15}$
pico	p	trillionth	$10^{-12}$
nano	n	billionth	$10^{-9}$
micro	$\mu$	millionth	$10^{-6}$
milli	m	thousandth	$10^{-3}$
centi	c	hundredth	$10^{-2}$
deci	d	tenth	$10^{-1}$
deka	da	ten	$10^1$
hecto	h	hundred	$10^2$
kilo	k	thousand	$10^3$
mega	M	million	$10^6$
giga	G	billion	$10^9$
tera	T	trillion	$10^{12}$
peta	P	quadrillion	$10^{15}$

When you are given a variable with a prefix you must convert it into its numerical equivalent before you use it in an equation.

*Always start by replacing the prefix symbol with its equivalent multiplier.*

*For example:  $0.16 \mu\text{A} = 0.16 \times 10^{-6} \text{A}$*

*$10 \text{ns} = 10 \times 10^{-9} \text{s}$*

*Replace the prefix with the equivalent multiplier:*

*e.g.  $12.0 \text{mm} = 12.0 \times 10^{-3} \text{m} = 1.2 \times 10^{-2} \text{m}$*

1.  $1.4 \text{ kW} =$

2.  $10 \mu\text{C} =$

3.  $24 \text{ cm} =$

4. 340 MW =

5. 46 pF =

6. 0.03 mA =

Express these with an appropriate prefix (Use of the ENG button on your calculator can help you)

e.g.  $5.6 \times 10^{-11} \text{m} = 56 \times 10^{-12} \text{m} = 56 \text{pm}$

7.  $6.4 \times 10^{-3} \text{W} =$

8.  $82.3 \times 10^8 \text{m} =$

9.  $2.0 \times 10^{-5} \text{m} =$

10.  $5.12 \times 10^8 \text{bytes} =$

11.  $1000 \times 10^{-9} \text{A} =$

12.  $0.520 \times 10^4 \Omega =$

Change the prefix to the one required (using the ENG and SHIFT+ENG buttons on your calculator can help you)

13. 0.0063 kA = mA

14. 300 Mms<sup>-1</sup> = Gms<sup>-1</sup>

15.  $6 \times 10^3 \text{km}$  = Mm

16. 0.0023 THz = MHz

$$17. 657,000 \mu\text{m} = \text{cm}$$

$$18. 2,112,000\mu\text{s} = \text{ms}$$

$$19. 0.0083\text{GBytes} = \text{kBytes}$$

## 2. Powers of 10

Calculate the following using the EXP or  $\times 10^x$  button on your calculator.

$$1. 8 \times 10^3 \times 9 \times 10^6 =$$

$$2. 5 \times 10^{-3} \times 4 \times 10^{-2} =$$

$$3. \frac{3.00 \times 10^8}{550 \times 10^{-9}} =$$

$$4. \frac{5.5 \times 10^3}{4.7 \times 10^6} =$$

$$5. \frac{(42 \times 10^3)^2}{384 \times 10^6} =$$

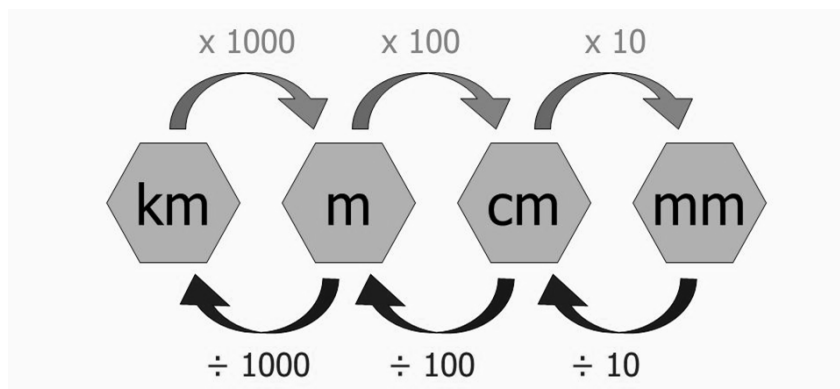
$$6. \frac{6.63 \times 10^{-34} \times 3.00 \times 10^8}{720 \times 10^{-9}} =$$

$$7. \frac{4.92 \times 10^{-7} \times 5000 \times 10^{-3}}{\pi \times (5.00 \times 10^{-5})^2} =$$

$$8. \frac{196 \times 10^{-3} \times 80 \times 10^{-2}}{\pi \times (5.0 \times 10^{-3})^2 \times 6.0 \times 10^{-3}} =$$

### 3. Conversions

Remember to raise the conversion factor to the same power as the units.



1.  $1000\text{cm}^2 = \text{m}^2$
2.  $5\text{mm}^2 = \text{m}^2$
3.  $5550\text{mm}^2 = \text{m}^2$
4.  $15\text{cm}^3 = \text{m}^3$
5.  $7000\text{cm}^3 = \text{m}^3$
6.  $11\text{mm}^3 = \text{m}^3$
7.  $6 \times 10^{-4} \text{ } ^\circ\text{Cs}^{-1} = \text{ } ^\circ\text{C per day}$
8.  $1000\text{kgm}^{-3} = \text{kgcm}^{-3}$
9.  $2 \times 10^{24} \text{ per m}^3 = \text{per mm}^3$
10.  $2\text{kg per min.} = \text{gs}^{-1}$

#### 4. Change of subject / Rearranging Equations

Formula	Rearrangements		
$\rho = \frac{m}{V}$	$V =$	$m =$	
$F = kx$	$k =$	$x =$	
$\varepsilon = \frac{x}{l}$	$l =$	$x =$	
$E = \frac{\sigma}{\varepsilon}$	$\varepsilon =$	$\sigma =$	
$G = \frac{\sigma A}{l}$	$A =$	$\sigma =$	$l =$
$R = \frac{\rho L}{A}$	$A =$	$\rho =$	$l =$
$\rho = \frac{1}{\sigma}$	$\sigma =$		
$v = u + at$	$u =$	$a =$	$t =$
$hf = \phi + E_k$	$f =$	$\phi =$	$E_k =$
$E_s = \frac{1}{2}k\Delta l^2$	$\Delta l =$		$k =$
$v^2 = u^2 + 2as$	$u =$	$a =$	$s =$



Express the following quantities in terms of their base units:

7. Force,  $F$  in Newtons(N), given that  $F = ma$
  
8. Work done,  $W$  in Joules(J), given that  $W = F \times s$
  
9. Power,  $P$  in Watts(W), given that  $P = \frac{W}{t}$
  
10. Momentum,  $p$ , given that  $F = \frac{p}{t}$
  
11. Density,  $\rho$ , given that  $\rho = \frac{m}{V}$
  
12. Angular frequency,  $\omega$ , given that  $\omega = \frac{2\pi}{T}$
  
13. The gravitational constant  $G$ , given that  $g = \frac{GM}{r^2}$

## 7. Density Calculations

1. A block of aluminium is 160kg and has a volume of 0.06 m<sup>3</sup>. What is the density of aluminium?

.....kgm<sup>-3</sup>

2. A block of material has a volume of  $32,000 \text{ cm}^3$  and a density of  $11 \times 10^3 \text{ kgm}^{-3}$ . What is the mass of the block?

.....kg

3. The density of gold is  $19300 \text{ kgm}^{-3}$ . The largest gold bar in the world measures 0.45cm by 170mm by 170mm, what is it's mass?

.....kg

4. What is the side length of a block of steel of mass 60kg and density  $7.7 \times 10^3 \text{ kgm}^{-3}$ .

.....m

5. Mercury is a liquid metal with a density of  $13.6 \text{ gcm}^{-3}$ . What is the mass of  $200 \text{ cm}^3$  of mercury?

.....kg

6. The volume of the Earth  $1.1 \times 10^{21} \text{ m}^3$ . Its mass is  $6.0 \times 10^{24} \text{ kg}$ . What is its average density?

..... $\text{kgm}^{-3}$

7. The hydrogen nucleus consists of a proton of mass  $1.7 \times 10^{-27} \text{ kg}$  and a radius of 1fm. What is the density of the hydrogen nucleus?

..... $\text{kgm}^{-3}$

8. A rectangular cuboid sample of brass(of density  $8200 \text{ kg/m}^3$ ) has a mass of 492g. It is 100mm long by 2cm wide. What is its height?

.....m

9. The density of iron is  $7800 \text{ kgm}^{-3}$ . Calculate the mass of an iron wire of length 2.5m and cross sectional area  $0.82 \text{ mm}^2$

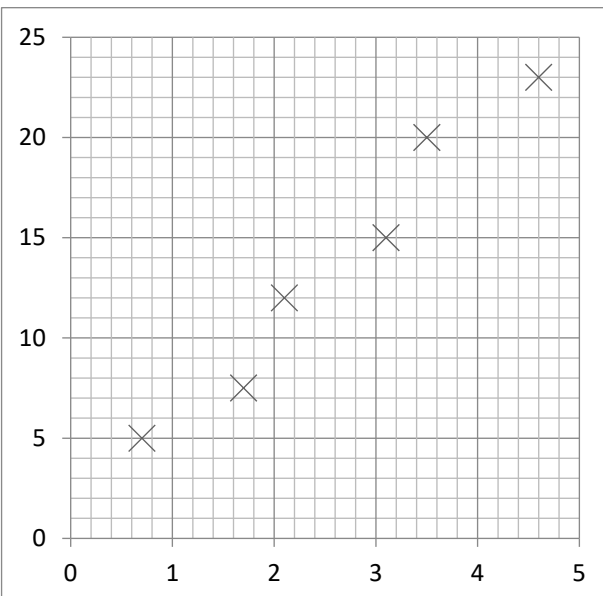
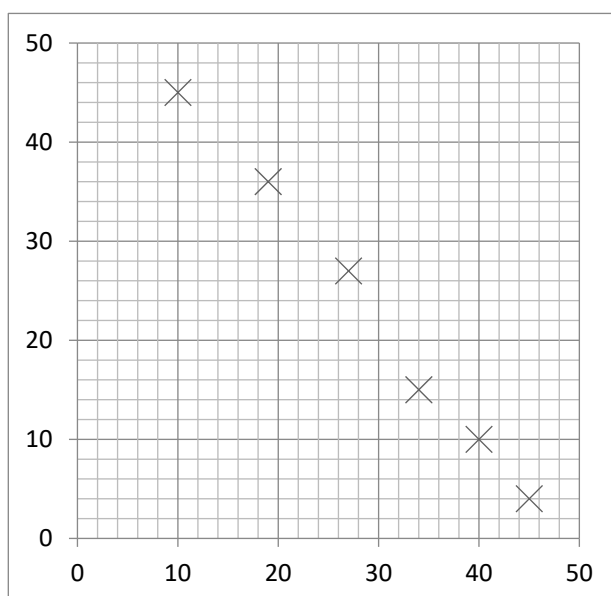
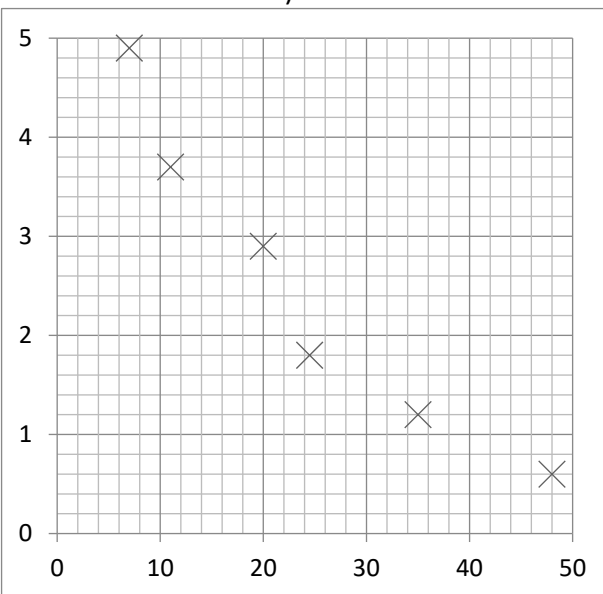
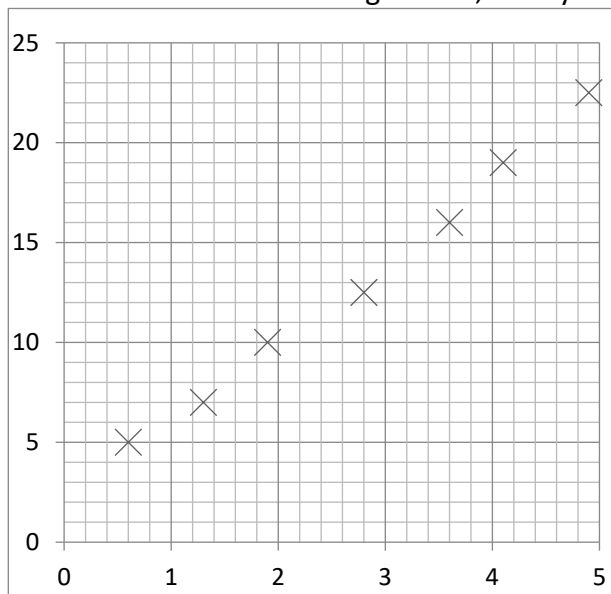
.....kg

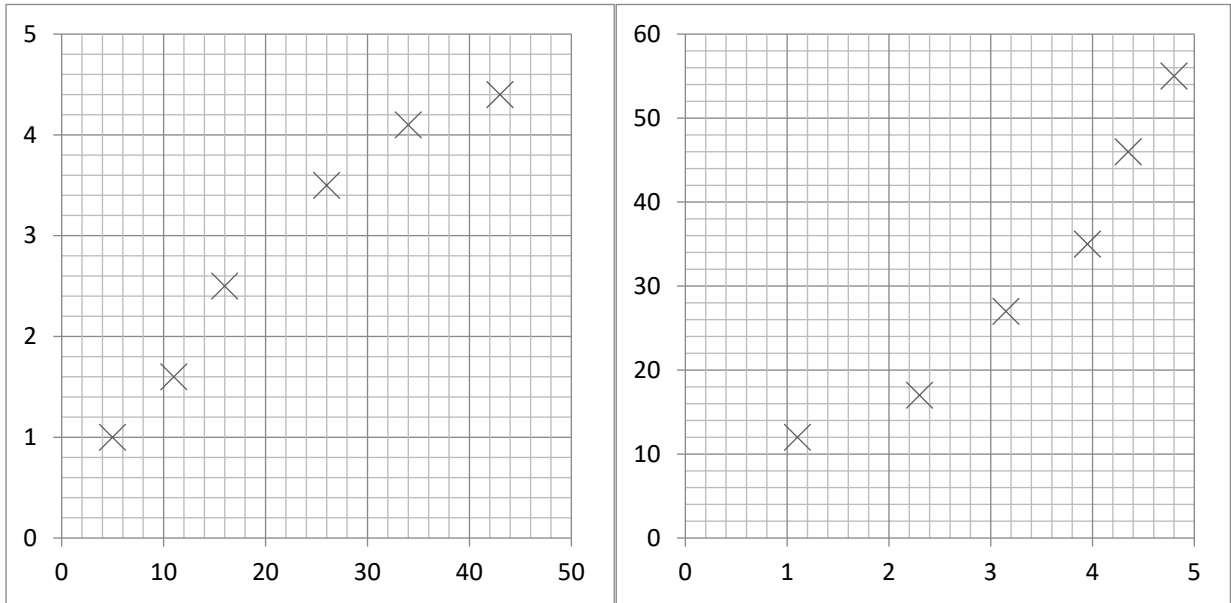
10.  $50 \text{ cm}^3$  of a material weighs 43g. Calculate the density of the material in  $\text{kgm}^{-3}$ .

..... $\text{kgm}^{-3}$

## 9. Graph Skills

1. draw a line of best fit on each of these graphs (think about the shape the data implies- lines of best fit do not have to be straight lines, if they are a rule should be used).





2. Complete the table identifying the quantities from the equation of a straight line, as in the example)

Equation	Graph	Rearranged Equation	Gradient	Intercept
$y = mx + c$	y plotted on the y axis	$y = mx + c$	$m$	$c$
	x plotted on the x axis			
$V = IR$	y axis = $V$	$V = RI$	$R$	$0$
	x axis = $I$			
$I = \frac{Q}{t}$	y axis = $t$			
	x axis = $Q$			
$\rho = \frac{RA}{l}$	y axis = $l$			
	x axis = $R$			
$\varepsilon = V + Ir$	y axis = $V$			
	x axis = $I$			
$E = VIt$	y axis = $E/t$			
	x axis = $V$			
$hf = \phi + E_K$	y axis = $E_K$			
	x axis = $f$			

$\lambda = \frac{h}{mv}$	y axis = $1/v$			
	x axis = $m$			
$E_p = mgh$	y axis = $mg$			
	x axis = $E_p$			
$E = \frac{1}{2}Fe$	y axis = $e$			
	x axis = $1/F$			
$c = f\lambda$	y axis = $1/\lambda$			
	x axis = $f$			
$v = u + at$	y axis = $a$			
	x axis = $1/t$			
$v^2 = u^2 + 2as$	y axis = $v^2$			
	x axis = $s$			
$s = \frac{(u + v)}{2}t$	y axis = $v$			
	x axis = $s$			
$w = \frac{\lambda D}{s}$	y axis = $\lambda$			
	x axis = $w$			